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ORIGINAL ARTICLE

Reduce antibacterial activity against ceftriaxone resist Escherichia coli isolated from urinary tract infections by modification of ceftriaxone

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Abstract

This study focused on clarifying the problem of antibiotic resistance in harmful bacteria and proposed a solution by utilizing the advantages of schiff base synthesis. Seventy-five urine samples were procured from a Baghdad, Iraq medical facility, forming the study's basis. Through a chemical reaction involving ceftriaxone and specific aldehydes, namely 4-hydroxybenzaldehyde and 4-ethoxybenzaldehyde, two novel Schiff base derivatives, labeled as compounds A and B, were synthesized and subsequently characterized using Fourier-transform infrared spectroscopy (FTIR). Intriguingly, conventional aldehyde derivatives and ceftriaxone exhibited limited efficacy in inhibiting Escherichia coli (E. coli) growth, as evidenced by reduced inhibition zones. In contrast, the newly synthesized Schiff base derivatives demonstrated enhanced antimicrobial activity, significantly inhibiting bacterial growth with zones measuring 21 mm for compound A and 18 mm for compound B. These findings underscore the potential of Schiff base synthesis as a promising avenue for developing effective strategies against antibiotic-resistant bacterial strains, highlighting the importance of innovative approaches in combating microbial resistance. ©2024 ijrei.com. All rights reserved

1. Introduction

Antimicrobial resistance, primarily caused by bacteria, is a prominent global public health problem [1]. The increasing antibiotic resistance in bacteria that are significant human pathogens, along with the transmission of resistance from the confined setting of hospitals to wider communities, poses a severe threat to public health [2]. Ceftriaxone is anticipated to exhibit efficacy against many gram-positive and gramnegative microorganisms [3].

Developing countries encounter unique challenges, from selfprescribing antibiotics to inadequate sanitary conditions, including those found in hospitals [4]. These conditions

Corresponding author: Marwa Mohssen Khudair Email Address: marwa.gen13@yahoo.com https://doi.org/10.36037/IJREI.2024.8103 contribute to the emergence and spread of specific multiresistant pathogens uncommon in developed countries. Furthermore, more research is needed into new antibiotics to combat these particular pathogens [5]. Moreover, it is crucial to acknowledge that these idiosyncrasies can readily transcend national boundaries in the current era of globalization, thereby impacting resistance patterns worldwide [6]. Different microorganisms like bacteria, fungi, and actinomycetes produce chemical compounds known as antibiotics [7], also known as antibacterial or chemotherapeutic medications [8]. These compounds can impede or eradicate other microorganisms. Antibiotics are chemically created medicines

that are not biologically derived, despite being called "antibiotics." А more appropriate designation is antibacterial/chemotherapeutic agents [9]. Ceftriaxone, a cephalosporin of the third-generation antibiotic, is commonly employed for the treatment of invasive infections caused by Enterobacteriaceae [10], including Escherichia coli. The rising global prevalence of antimicrobial resistance (AMR) in Enterobacteriaceae is leading to higher rates of patient illness and death [11], increased expenses in healthcare, and more reliance on medicines that are used as a last resort. The World Health Organization has acknowledged ceftriaxone-resistant Enterobacteriaceae as a crucial priority pathogen. Extendedspectrum B-lactamases (ESBLs) and AmpC B-lactamases are the primary mechanisms of ceftriaxone resistance in Enterobacteriaceae [12]. The genes responsible for generating these enzymes can be horizontally transmitted from one bacterium to another, which facilitates the dissemination of resistance [13].

This study aimed to determine how well ceftriaxone and schiff bases kill *E. coli* strains resistant to Ceftriaxone.

2. Methods

2.1 Synthesis of Schiff base derivatives (A and B)

For ceftriaxone modification as Schiff base derivatives, dissolve 4-hydroxybenzaldehyde (1.22 gm, 0.01 mol) and 4ethoxybenzaldehyde (1.5 gm, 0.0 mol), respectively, in 20 ml of absolute ethanol. Then, the ceftriaxone (5.54 gm, 0.01 mol) was added to these solutions and refluxed for 3 h at 70 °C with stirring. After 4 hours, the solutions were cooled to room temperature. The yellow precipitates were filtered, washed with ethanol, and dried in a vacuum [14]. The structure of these derivatives is shown in Fig .1.



Figure 1: Structures of Schiff base (A and B).

2.2 Antimicrobial Activity of Schiff base derivatives (A and B)

The *E. coli* bacteria were cultivated on Muller-Hinton dishes via a sterile loop and streaking techniques, starting with the broth [15]. Subsequently, a single well was generated in the agar. Each well received 100 μ l of the relevant dilution of Schiff base derivatives A and B (ceftriaxone and aldehyde derivatives). The dish was hermetically sealed and incubated at 37 °C for the night to be examined the following day.

2.3 Dilution of Ceftriaxone

With a micropipette, 100 μ l of the stock solution (made by dissolving 1 gm of the antibiotic in 10 ml of distilled water) was added to the distilled water (900 μ l) in an Eppendorf tube to produce a solution (1). Three dilutions were prepared using the serial dilution approach [16].

2.4 Resistance to Ceftriaxone

E. coli bacteria were cultured in a media (Muller-Hinton), and a ceftriaxone disk (40 μ g) was placed onto the agar. The bacteria grew exponentially overnight [17].

2.5 Experimental of Docking

The protein data bank (PDB code: 1MOQ) was used to get the crystallographic 3D structure of glucose-6-phosphate synthase (GP6 synthase). The enzyme structure was put through energy minimization using Swiss Protein Viewer (version 4.1). Several preparatory stages were executed, resulting in the removal of molecules of water and co-crystallized compounds. Subsequently, the structure was repaired, and polar hydrogens were included. The compounds produced were shown using ChemDraw Ultra (version 18.0) and stored in a sdf format file.

The energy minimization was performed via Chem3D (version 18.0), then converted to a pdb file format by OpenBabel software. We used Autodock Tools (version 1.5.6) [18, 19], Eberhardt, Santos-Martins et al. 2021) to do the docking process. We saved the results in pdbqt file format after minimizing the energies of both enzymes and ligands. The molecular docking was conducted using Autodock Vina with the AutoGrid program. The grid box's dimensions were 40 x 40 x 40, with a grid center of 21.067 x 20.007 x 12.729, representing the x, y, and z dimensions, respectively. The docking settings were configured with the default values,

resulting in the generation of nine conformations. Discovery Studio was used to generate display pictures [20].

3. Results and Discussion

FTIR (cm⁻¹) of Schiff base derivative (A): the broad band of hydroxyl group appeared at 3359, C-H aromatic at 3089, C-H aliphatic appeared at 2934 and 2882, and the carbonyl group of carboxylic acid and amide appeared at 1718 and 1668, respectively. The amine group of Schiff base at 1642 and C-C aromatic at 1592 [21], as shown in Fig. 2.



Figure 2: FTIR for derivative A.

FTIR (cm⁻¹) of schiff base derivative (B): the broad band of hydroxyl group appeared at 3340, C-H aromatic at 3046, C-H aliphatic appeared at 2940 and 2880, and the carbonyl group of carboxylic acid and amide appeared at 1717 and 1690, respectively. The imine group of Schiff base at 1642 and C-C aromatic at 1593 [22], as shown in Fig. 3.

3.1 Discussion of Docking

Docking models are crucial for investigating the binding mechanisms of ligands to a target molecule. The significance of glucosamine-6-phosphate synthase (GP6 synthase) in microbial cell wall formation has garnered attention from numerous researchers [23]. In the first step of hexosamine biosynthesis, the enzyme speeds up the process by changing fructose-6-phosphate into glucosamine-6-phosphate. GlcN-6-P is a building block for uridine diphosphate N-acetyl glucosamine (UDP-NAG), which is an important part of the peptidoglycan layer in the cell walls of microbes [24]. The docking profiles of ceftriaxone and its derivative are shown in Fig. 4–7. The 3D conformations of both ligands show multiple interactions, most notably hydrogen bonding, in addition to other interactions such as van deer Waals and pi-alkyl. The binding affinities for both ligands were comparable, with computed values of -6.9 kcal/mol and -7.3 kcal/mol for ceftriaxone and ethoxy-ceftriaxone, respectively. Experimental findings further support the findings that both ligands exhibit anti-bacterial activity.



Figure 3: FTIR for derivative B

3.2 Aldehyde Derivatives (4-hydroxybenzaldehyde and 4ethoxybenzaldehyde) Sensitivity Test

0.05 μ g/ μ l Schiff base derivatives (A and B) were added to wells on Muller-Hinton agar. Subsequently, the plates were infected with an overnight culture of bacteria. The next day, the data and the antibacterial activity of Schiff base derivatives (A and B) against *E. coli* bacteria were looked at. The results showed 6 and 5 mm of inhibition zones, respectively. The Dens Check machine was employed to standardize the first Kahn tube to the McFarland standard (1.5 x 108 CFU/ml) [25].



Figure 4: Docking profile 2D of ceftriaxone against GP6 synthase.



Figure 5: Docking profile 3D of ceftriaxone against GP6 synthase.

3.3 Sensitivity Test of Ceftriaxone

E. coli was cultured overnight on Muller-Hinton agar in the presence of a ceftriaxone disk (40 μ g). Upon analysis overnight, the results showed 10 mm of inhibition zones of the growth zone [26].

3.4 Ceftriaxone minimum inhibitory concentration test

Ceftriaxone, used at a dose of 80 μ g/ μ l, clearly inhibits the investigated isolates. The diameter of the inhibition zone in comparison to the antibiotic disc determines the degree of inhibition. However, no antibacterial effect was observed when the antibiotic was tested at different concentrations of 10, 5, and 1 μ g/ μ l, demonstrating the dilutions' effect on *E. coli*.

The antibacterial properties of the Schiff base exhibited significant inhibitory activity, resulting in inhibition zones measuring 21 and 18 mm, respectively. In comparison, 4-hydroxybenzaldehyde and 4-ethoxybenzaldehyde, when applied individually, only produced inhibition zones measuring 6 and 5 mm, respectively. Fig. 8 displays the inhibition areas observed due to combining ceftriaxone with Schiff base (A) [27].



Figure 6: Docking profile 2D of Schiff base derivative (B) against GP6 synthase.



Figure 7: Docking profile 3D of Schiff base derivative (B) against GP6 synthase.

Based on the results obtained, the Schiff base derivatives of Ceftriaxone demonstrate a higher capacity to hinder the development of *E. coli* on Muller-Hinton agar. As previously stated, the absence of any inhibitory zone indicates that *E. coli* exhibited resistance to the presence of unbound

Ceftriaxone. The current study also shows that free aldehyde derivatives can prevent *E. coli* from growing near the aldehyde derivatives well, creating a 6- and 5-mm wide inhibition zone [28].

The combination of aldehyde derivatives with Ceftriaxone has resulted in enhanced bactericidal activity of the Schiff bases (ceftriaxone and aldehyde derivatives), with killing zones of 21 and 18 mm for the initial dosage of aldehyde derivatives and the first dilution of Ceftriaxone, respectively. Table 1 illustrates the different concentrations of Schiff bases (A and B) along with their corresponding killing zone diameter and the percentage increase for each concentration.



Figure 8: Schiff base derivative (A) on Muller–Hinton agar.

Studies have demonstrated that Schiff base derivatives of ceftriaxone modification identified with antibiotics may increase the levels of antibiotics at the location where bacteria and antibiotics come into contact and facilitate the binding of antibiotics to microorganisms.

Juan A. et al., Synthesized a tridentate schiff base derived from cephalexin. The present finding confirms that ceftriaxone effectively produces synergistic antibacterial actions against several types of bacteria. In addition, cephalexin, along with other cephalosporins, has shown increased antibacterial efficacy against bacteria when combined with aldehyde derivatives [29], which agrees with our study.

Abdelbaset A., Schiff bases are derived from cephalexin. In these derivatives, an amino group available in the drug substances was allowed to react with two derivatives of aldehyde separately to obtain Schiff bases [30]. This study agreed with ours.

Table 1: Inhibition zone of E. coli.

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	No. of comparison	4-hydroxybenzaldehyde	4-ethoxybenzaldehyde	Schiff base (A)	Schiff base (B)
	Comparison #1	6	5	21	18
	Comparison #2			20	16
	Comparison #3			18	15

4. Conclusions

In this study, we successfully synthesized two novel Schiff base derivatives, namely compounds A and B, through the chemical reaction involving ceftriaxone and specific 4-hydroxybenzaldehyde aldehydes: and 4ethoxybenzaldehyde. To validate their molecular composition and structure, these derivatives underwent characterization using Fourier-transform infrared spectroscopy (FTIR). Intriguingly, when these Schiff base derivatives were combined with ceftriaxone, they manifested a synergistic effect. This synergism notably amplified the inhibition zone, indicating enhanced efficacy against the targeted agents or organisms. Despite these promising observations, а comprehensive understanding of this synergistic interaction's underlying mechanisms and full potential still needs to be discovered. Therefore, further intricate studies and analyses are imperative to elucidate the precise nature of this synergistic impact. Such in-depth investigations will validate our initial findings and pave the way for potential applications and advancements in effectively combating relevant challenges.

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